

Detection of Sodium Halides in Aqueous Systems Using Ultra-Thin Hydrogel Layers on a Quartz Crystal Microbalance

Jackie A. Shepard, Venkat R. Bhethanabotla, and Ryan G. Toomey

ChE. 9

Sensors Research Laboratory, Clean Energy Research Center - University of South Florida Department of Chemical Engineering, Tampa, FL 33620-5350

Stimuli-Responsive Hydrogel Networks

Motivation

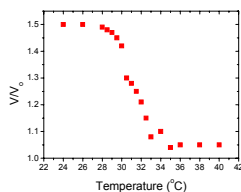
- Stimuli-responsive hydrogel networks carry a potential role in biological systems, foods, cosmetics, pharmaceuticals, and other industries, and are not fully understood.
- The effect of salts on hydrogel networks that expand and contract in one-dimension has never been demonstrated.

Our goal is to develop a set of "intelligent" hydrogel materials that recognize and communicate ion binding to demonstrate the effect of salt stimuli on one-dimensional transition networks.

Concept

What is an LCST network? It is a polymer network that undergoes a reversible phase transition at a well-defined temperature. The temperature is determined by the sum of hydrogen bonding, ionic, and van der Waals interactions in the network.

Swelling Behavior:

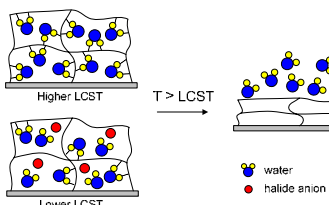


T = 24 °C
Swollen State (hydrophilic)

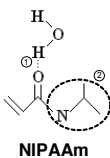


T = 35 °C
Collapsed State (hydrophobic)

Why use an LCST network? The transition temperature is an effective gauge of the "nanoscale" environment.



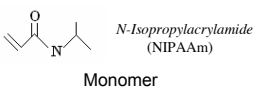
- Ions destroy the water structure around polymers.
- Ions interact with polymer chains.
- LCST is strongly dependent on anions: $I < Br < Cl < F$



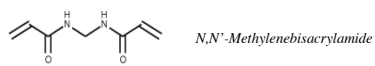
NIPAAm

Hydrogel Preparation

The hydrogel networks are fabricated through UV illumination using free radical polymerization *in situ*:



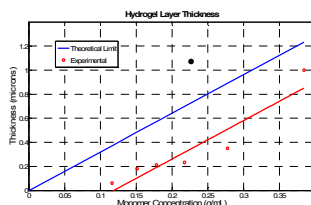
Monomer



Cross-linker

Thickness Information

- Dry hydrogel thickness is controlled through the concentration of monomer added prior to polymerization.
- Polymerization process carried out in an inert gas atmosphere:



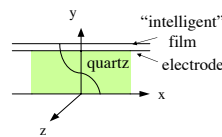
- Theoretical thickness calculation:

$$\text{Thickness} = \frac{\text{concentration} \times \text{amount deposited}}{\text{density} \times \text{area of deposition}}$$

- The hydrogel thickness provides different levels of ion sensitivity.

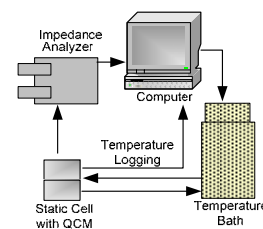
Quartz Crystal Microbalance (QCM) Principles

- Principle:** The quartz crystal vibrates at its fundamental resonance frequency.
- Changes in the mechanical stress at the quartz surface induces a measurable shift in its fundamental resonance frequency.



- The targeted analyte will be "sensed" based on the temperature at which the hydrogel collapses.

Experimental Setup: water with the target sodium halide is contained within the static cell. The temperature bath provides heat for hydrogel collapse.

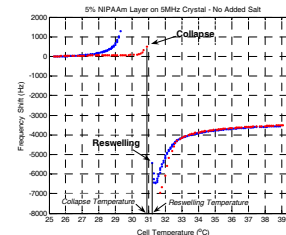


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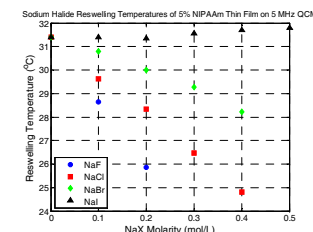
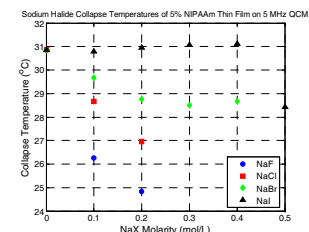
Results

- Frequency response of collapse and reswelling of NIPAAm thin film with no added salt:



- Continuous transition
- Transition marked by a highly ordered system to a disordered one
- Transition is distinguishable by a frequency shift of 3 to 4 kHz
- Three runs per NaX per concentration were averaged to find the transition temperatures
- Average transition temperatures varied by 0.1 to 0.2 °C in most cases

- Transition temperatures of NIPAAm thin film with various concentrations of different sodium halides:



Future Directions

- Turbidity measurements to compare one-dimensional transitions to three-dimensional swelling transitions.
- Investigation of the LCST behavior of the NIPAAm in conjunction with metalloptides for heavy metal ion detection in potable water.
- Antibody/Antigen incorporation into the NIPAAm network for earlier detection of specific diseases.

Conclusions

- Detection of sodium halides is based on the signal response of the QCM to a change in swelling and viscoelastic properties of the hydrogel network upon ion interaction. The fabrication of ultra-thin pNIPAAm hydrogel matrices has been demonstrated.
- Detection would offer reusability and reproducibility as shown by its frequency response recovery upon heating to its LCST.
- Sodium halide concentration within the hydrogel matrix can be "transduced" by the QCM into a readable signal.



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